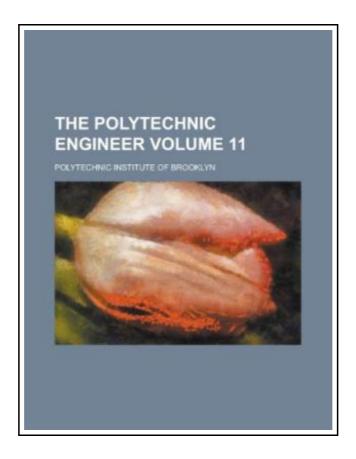
# The Polytechnic engineer Volume 11



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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 48 pages. Dimensions: 9.7in. x 7.4in. x 0.1in.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1911 edition. Excerpt: . . . N, O and F can be calculated from the assumption that the atomic volume of each of these elementary gases is roughly proportional to its atomic weight, the respective specific gravities resulting being 1. 44, 1. 38, and 1. 15. By the employment of a similar method, the atomic volumes of liquid He and Ne are found to be about 5. 1 and 17. 2, respectively. Type line He Ne can now be drawn, the negative slope of N P corrected, line H F fixed with greater accuracy, and the true slant of O S determined. The precession of the type lines along corresponding group lines can be traced completely; and it will be found that the lines of the typical elements show a tendency to revolve clockwise in the order of their groups, from an angle of about 60 for the H F line to about 30 for that of O S. After this, the rotation is continued by the lines of the corresponding chemical families reading (in the Periodic Table), however, from left to right from the CI Br I group to that of Mn and Sm, and, simultaneously from right to left from the Se Te group back to that of Fe, Ru and Os. The slope of the Argon group would doubtless not be so steep if the density of the solid gases were determined. Likewise the densities of solid chlorine and bromine should determine a more consistent slant than...



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